

Compound	R ₁	R ₂
1	H	1'-α-(3-amino-2,3,6-trideoxy-L-lyxosyl)
2	OH	1'-α-(3-amino-2,3,6-trideoxy-L-lyxosyl)
3	H	H
4	OH	H

Table 1. Carbon chemical shifts, Δ (ppm) and labelling pattern found for the aliphatic sidechains of 3 and 4

Carbon No.	Daunomycinone (3) unlabelled	labelled	Adriamycinone (4) unlabelled	labelled
7	60.5		60.4	
8	35.6		36.2	
9		76.2		76.2
10	31.9		32.5	
13	211.8		213.6	
14	24.4		64.1	

Δ run in DMSO-*d*₆, results are reported relative to TMS using the relationship δ(TMS) = δ(DMSO-*d*₆) - 39.5 ppm.

one (4), respectively. The ¹³C-NMR chemical shift assignments for the aliphatic sidechains of both aglycones are summarized in Table 1 and are based upon literature values [4].

A comparison of the intensities of the appropriate signals in spectra of the labelled and unlabelled compounds revealed that, in both cases, the quaternary aliphatic ring carbon (C-9, 76.2 ppm) was enriched *ca* 10-fold. This result constitutes direct support for the postulate that a propionate (C-15, C-14 and C-9) 'starter' is directly involved in the biosynthesis of both daunomycin and adriamycin in this organism. A similar result has been observed for the related metabolite ξ-pyrrromycinone using ¹⁴C-tracer studies [5].

EXPERIMENTAL

Culture. The source of *Streptomyces peucetius* var. *caesius* (IMRU No. 3920) and conditions used for its maintenance have been described [2].

Isotopic incorporation studies. Submerged production cultures of *S. peucetius* var. *caesius* were grown at 28° in 300 ml flasks containing 40 ml medium and stirred on a shaker rotating at 220 rpm with a 7 cm radius [2]. 48 hr after inoculation, 40 mg sodium propionate [¹⁻¹³C] (90 atoms % ¹³C, Merck) was

added to each flask. After a further 72 hr incubation the whole culture was exhaustively extracted as previously described [2]. Mild acid hydrolysis of crude 1 and 2 with 0.1 N HCl at 100° for 20 min gave a mixture of dark red coloured aglycones, 3 and 4. Each metabolite was separated and purified by repeated PLC on Si gel developed with CHCl₃-Me₂CO (4:1). PFT-¹³C-NMR spectra were recorded on a Varian XL100 (25.2 MHz) spectrometer using DMSO-*d*₆ as solvent.

Acknowledgement—The award of an N.I.H. Visiting Fellowship to G. J. S. is acknowledged.

REFERENCES

1. Arcamone, F., Franceschi, G., Orezzi, P., Cassinelli, G., Barbieri, W. and Mondelli R. (1964) *J. Am. Chem. Soc.* **86**, 5334.
2. Arcamone, F., Cassinelli, G., Fantini, G., Grien, A., Orezzi, P., Pol, C. and Spalla, C. (1969) *Biotechnol. Bioeng.* **11**, 1101.
3. Paulick, R. C., Casey, M. L. and Whitlock, H. W. (1976) *J. Am. Chem. Soc.* **98**, 3370.
4. Arnone, A., Fronza, G., Mondelli, R. and Vigevari, A., (1976) *Tetrahedron Letters* **37**, 3349.
5. Ollis, W. D., Sutherland, I. O., Codner, R. C., Gordon, J. J. and Miller, G. A. (1960) *Proc. Chem. Soc. (London)* 347.

NEUE TREMETON-DERIVATE AUS *DORINICUM MACROPHYLLUM*

FERDINAND BOHLMANN und MICHAEL GRENZ

Institut für Organische Chemie der Technischen Universität Berlin, Straße des 17. Juni 135, D-1000 Berlin 12, W. Germany

(Eingegangen am 23. Juni 1978)

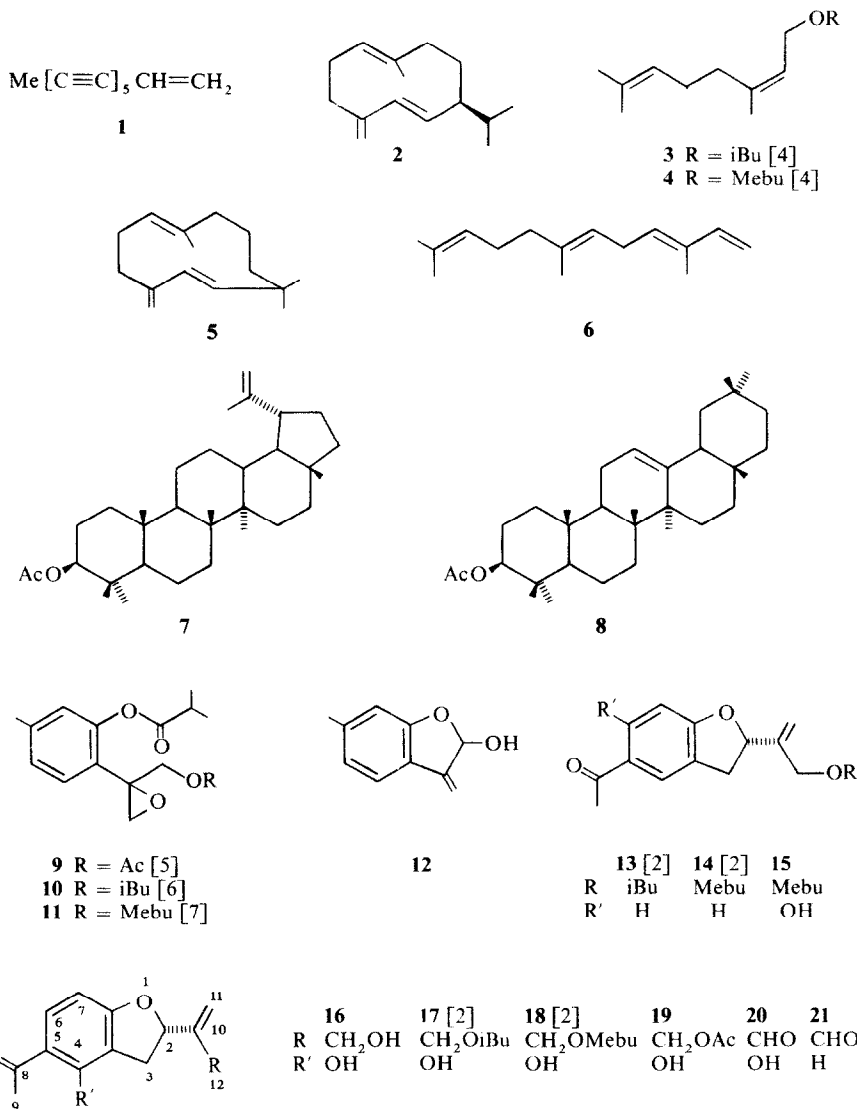
Key Word Index—*Doronicum macrophyllum*; Senecioneae; Compositae; new tremetone derivatives; benzofurans.

Bisher sind aus Vertretern der Gattung *Doronicum* (Tribus Senecioneae) neben dem weitverbreiteten Pentainen (1) [1] aus einer Art Thymol- und Tremeton-Derivate [2] isoliert worden.

Auch die Wurzeln von *Doronicum macrophyllum* enthalten neben 1–4 die Thymolderivate 9–11 und die Tremeton-Derivate 17 und 18. Daneben isoliert man drei weitere derartige Verbindungen, denen nach den spektroskopischen Daten die Konstitutionen 15, 16 und 21 zukommen (s. Tabelle 1).

Die oberirdischen Teile, aus denen bereits das Pyrrolizidin-Alkaloid Doronin isoliert wurde [3], ergeben neben 1, 2–8, 12 und 16–18 zwei weitere Tremeton-Derivate, denen die Konstitutionen 19 und 20 zukommen (s. Tabelle 1). Die Stellung der phenolischen OH-Gruppe ergibt sich aus der Lage und der Kopplung der aromatischen Protonen-Signale.

Die vorliegenden Ergebnisse geben keine neuen Anhaltspunkte über eine mögliche Umgruppierung dieser Gattung in eine andere Tribus. Ähnliche Tremeton-

Tabelle 1. ¹H-NMR-Spektren von **15**, **16**, und **19–21** (270 MHz, CDCl₃, TMS als innerer Standard)

	15	16	19	20	21
2-H	<i>dd</i> (<i>br</i>) 5.37	<i>dd</i> (<i>br</i>) 5.48	<i>dd</i> 5.44	<i>dd</i> (<i>br</i>) 5.68	<i>dd</i> (<i>br</i>) 5.64
3-H	<i>dd</i> 3.28	<i>dd</i> 3.41	—	<i>dd</i> 3.60	<i>dd</i> 3.68
3'-H	<i>dd</i> 3.11	<i>dd</i> 3.10	—	<i>dd</i> 2.88	<i>dd</i> 2.94
4-H	<i>t</i> 7.51	—	—	—	<i>dt</i> 7.82
6-H	—	<i>d</i> 7.61	<i>d</i> 7.62	<i>d</i> 7.64	<i>dd</i> 7.84
7-H	<i>s</i> 6.38	<i>d</i> 6.41	<i>d</i> 6.50	<i>d</i> 6.47	<i>d</i> 6.90
9-H	<i>s</i> 2.55	<i>s</i> 2.56	<i>s</i> 2.57	<i>s</i> 2.57	<i>s</i> 2.55
11-H	<i>s</i> (<i>br</i>) 5.34	} <i>s</i> (<i>br</i>) 5.28	<i>s</i> (<i>br</i>) 5.36	<i>d</i> 6.52	<i>d</i> 6.56
11'-H	<i>s</i> (<i>br</i>) 5.28		<i>s</i> (<i>br</i>) 5.30	<i>d</i> 6.18	<i>d</i> 6.19
	<i>f</i> <i>d</i> 4.71				
12-H	<i>d</i> 4.64	ABq 4.27	<i>s</i> 4.70	<i>s</i> 9.66	<i>s</i> 9.66
OH	<i>s</i> 12.97	<i>s</i> 12.74	<i>s</i> 12.75	<i>s</i> 12.70	—
OCOR	<i>tq</i> 2.39	—	<i>s</i> 2.06	—	—
	<i>dq</i> 1.67	—	—	—	—
	<i>dq</i> 1.48	—	—	—	—
	<i>t</i> 0.90	—	—	—	—
	<i>d</i> 1.15	—	—	—	—

$J(\text{Hz})$: 2,3 = 9.5; 2,3' = 7.5; 3,4 = 1; 4,6 = 1.5; 6,7 = 8; bei **15**: 12,12' = 12.5; 2',3₁' = 2,3₂' = 2',5' = 7; 3₁',3₂' = 14; 3',4' = 7; bei **20/21**: 2,11 = 1.5; 2,11' = 1.

Derivate findet man auch in mehreren *Senecio*-Arten, so daß zusammen mit dem Vorkommen von Pyrrolizidinalkaloiden die Eingruppierung in die Tribus Senecioneae gerechtfertigt ist.

EXPERIMENTELLES

IR: Beckman IR 9, CCl_4 ; $^1\text{H-NMR}$: Bruker WH 270; MS: Varian MAT 711, 70 eV. Direkt einlaß. Die frisch zerkleinerten Pflanzenteile (angezogen aus Samen vom Botanischen Garten Nijmegen) wurden mit Ether/Petrol 1:2 extrahiert und die erhaltenen Extrakte zunächst durch SC (Si gel Akt. St. II) und weiter durch mehrfache DC (Si gel GF 254) aufgetrennt. Als Laufmittel dienten Et_2O /Petrol-Gemische. Bereits bekannte Substanzen identifizierte man durch Vergleich der IR- und $^1\text{H-NMR}$ -Spektren mit denen authentischer Verbindungen.

1 kg Wurzeln ergaben 5 mg 1, 15 mg 2, 50 mg 3 und 4 (ca 3:1), 38 mg 10 und 11 (ca 4:1), 20 mg 9, 12 mg 15 (Et_2O /Petrol 1:1), 23 mg 21 (Et_2O /Petrol 1:1) 13 mg 16 (Et_2O /Petrol 1:1) und 150 mg 17 und 18 (ca 5:1). 3 kg oberirdische Teile lieferten 1 mg 1, 20 mg 2, 45 mg 5, 50 mg 6, 17 mg 3 und 4 (ca 3:1), 19 mg 12, 5 mg 19 (Et_2O /Petrol 1:1), 10 mg 20 (Et_2O /Petrol 1:1), 20 mg 17 und 18 (ca 4:1), 5 mg 16 und 280 mg 7 und 8 (ca 1:1).

6,12-Dihydroxytremeton-12-O-[2-methylbutyrat] (15). Farbloses Öl, IR: CO_2R 1740; PhCO (brückengebunden) 1640 cm^{-1} . MS: M^+ m/e 318.147 (6%) (ber. für $\text{C}_{18}\text{H}_{22}\text{O}_5$ 318.147); $-\text{C}_4\text{H}_9\text{CO}_2\text{H}$ 216 (26); $\text{C}_4\text{H}_9\text{CO}^+$ 85 (50); 85 $-\text{CO}$ 57 (100).

4,12-Dihydroxytremeton (16). Farbloses Öl, IR: OH 3620; PhCO (brückengebunden) 1645 cm^{-1} . MS: M^+ m/e 234.089 (56%) (ber. für $\text{C}_{13}\text{H}_{14}\text{O}_4$ 234.089); $-\text{H}_2\text{O}$ 216 (35); 216 $-\text{Me}$ 201 (26); C_3H_7^+ 43 (100).

4-Hydroxy-12-acetoxytremeton (19). Farbloses Öl, IR: OAc 1750, 1230; PhCO (brückengebunden) 1645 cm^{-1} . MS: M^+ m/e 276.100 (5%) (ber. für $\text{C}_{15}\text{H}_{16}\text{O}_5$ 276.100); $-\text{AcOH}$ 216 (38); 216 $-\text{Me}$ 201 (13); MeCO^+ 43 (100).

4-Hydroxy-12-oxotremeton (20). Farblose Kristalle, Schmp. 80° (Petrol) IR: $\text{C}=\text{CHO}$ 2740, 2700, 1695, 1620; PhCO (brückengebunden) 1650 cm^{-1} . MS: M^+ m/e 232.074 (100%) (ber. für $\text{C}_{13}\text{H}_{12}\text{O}_4$ 232.074); $-\text{Me}$ 217 (25); $-\text{CO}$ 204 (27); 176 $-\text{Me}$ 161 (79).

12-Oxotremeton (21). Farbloses Öl, IR: CHO 2700, 1695; PhCO 1685, 1600 cm^{-1} . MS: m/e 216.079 (100%) (ber. für $\text{C}_{13}\text{H}_{12}\text{O}_3$ 216.079); $-\text{Me}$ 201 (92); 201 $-\text{CO}$ 173 (21); 173 $-\text{CO}$ 145 (82).

Anerkennung—Der Deutschen Forschungsgemeinschaft danken wir für die Förderung dieser Arbeit.

LITERATUR

1. Bohlmann, F., Burkhardt, T. und Zdero, C. (1973) *Naturally Occurring Aketylenes*. Academic Press, London.
2. Bohlmann, F. und Zdero, C. (1970) *Tetrahedron Letters* 3575.
3. Alieva, S., Abdullaev, U., Telezhnetkaya, M. und Yumusov, S. (1976) *Khim. Prir. Soedin*, 194.
4. Bohlmann, F., Zdero, C. und Faass, U. (1973) *Chem. Ber.* 106, 2904.
5. Bohlmann, F. und Zdero, C. (1976) *Chem. Ber.* 109, 791.
6. Bohlmann, F. und Zdero, C. (1972) *Tetrahedron Letters* 2827.
7. Bohlmann, F., Jakupovic, J. und Lonitz, M. (1977) *Chem. Ber.* 110, 301.

A NEW CHROMONE GLUCOSIDE FROM *TECOMELLA UNDULATA*

V. K. GUJRAL, S. R. GUPTA* and K. S. VERMA

Department of Chemistry, University of Delhi, Delhi-110007, India

(Received 23 June 1978)

Key Word Index—*Tecomella undulata*; Bignoniaceae; 2-methyl-5,7-dihydroxychromone 7-O- β -D-glucopyranoside.

Previous investigations [1–4] of the genus *Tecomella* have revealed the presence of lapachol, dehydrotectol, *n*-hentriacontanol, *n*-heptacosane, *n*-nonacosane, tecomin and tecomelloside. We now report the isolation and structure determination of a new chromone glucoside from the bark of *T. undulata*.

EtOAc insoluble portion of the alcoholic extract of the bark (2 kg) was dissolved in a minimum amount of MeOH and dry ether was added with constant shaking. The precipitate thus obtained on column chromatography over Si gel using (CHCl_3 –MeOH, 4:1) as eluent yielded the new compound as an amorphous powder, 500 mg; mp $235\text{--}37^\circ$; $[\alpha]_D^{27} -169.0$ (c, 0.71, MeOH) (Found: C, 51.4; H, 5.1. $\text{C}_{16}\text{H}_{18}\text{O}_9 \cdot \text{H}_2\text{O}$ requires: C, 51.6; H, 5.4%; R_f : 0.54 (EMW, 100:16.5:13.5); 0.39 (CHCl_3 –MeOH, 9:1); 0.5 (EtOAc – EtCOMe – HCO_2H –

H_2O , 5:3:3:1); 0.33 (EtOAc –MeOH, 5:1). It gave brown ferric reaction and positive Molisch's test. IR(KBr) showed strong absorptions at 1660 cm^{-1} (chelated $\text{C}=\text{O}$) and 3450 cm^{-1} . MS exhibited M^+ 354 and an aglycone peak formed by the loss of a six carbon sugar. These observations coupled with UV spectrum ($\lambda_{\text{max}}^{\text{MeOH}}$ (log ϵ) 255(4.15), 285(3.85), 315(3.6); AlCl_3 , 255(sh), 302, 362; AlCl_3 –HCl, 255(sh), 302, 362; NaOAc, 255(sh), 341 nm) indicated it to be a chromone glycoside. It formed a pentaacetate, mp $129\text{--}30^\circ$ (Py/ Ac_2O); $[\alpha]_D^{22} -100.2$ (c, 0.2, CHCl_3). NMR (60 MHz, δ) (CDCl_3) 2.05(b, 12H, four alcoholic acetoxy), 2.32(s, 3H, phenolic acetoxy), 2.43(s, CH_3 at C-2), 6.84(d, $J = 2.5\text{ Hz}$, 1H at C-8), 6.71(d, $J = 2.5\text{ Hz}$, 1H at C-6), 6.12(s, 1H at C-3), 4.9–5.6(sugar protons).

Acid hydrolysis gave an aglycone, mp $280\text{--}81^\circ$, M^+ 192, $\text{C}_{10}\text{H}_8\text{O}_4$ and a sugar unit identified as glucose

*To whom correspondence should be addressed.